

13+ Entrance Assessment 2013 Science 1 hour

Candidate Name:

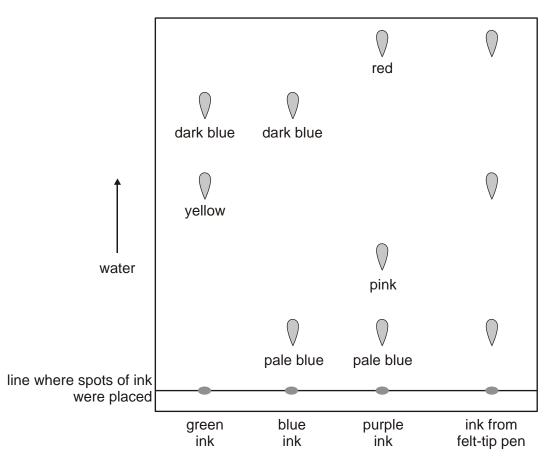
CHEMISTRY

1. Susie used chromatography to identify the coloured substances in the ink from a felt-tip pen.

She used:

- green ink
- blue ink
- purple ink
- ink from her felt-tip pen.

She used water as the solvent.



Look at the diagram above.

(a) (i) Which colours were present in the ink from the felt-tip pen?

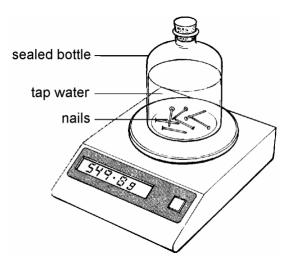
.....

(ii)	How many co	oloured substan	nces were there	in green ink?

		How can you tell?	
			1 mark
	(iii)	Susie placed the spots of ink on a line on the chromatography paper as shown in the diagram. To draw the line, Susie had to choose a felt-tip pen or a pencil.	
		Which one should she use?	
		Give the reason for your answer.	
			1 mark
(b)	When not v	e used water as the solvent in this experiment. n she repeated the experiment with a different set of pens, it did vork. hen used ethanol instead of water.	
	Sugg	est why the experiment worked with ethanol but not with water.	
			1 1
			1 mark

maximum 4 marks

2. An experiment was set up to investigate rusting. Some clean, shiny, iron nails were sealed in a glass bottle containing some tap water. The sealed bottle was then placed on a top-pan balance. The reading on the balance was 549.8 g.



The sealed bottle was left for one week. After one week the nails were rusty.

(a) (i)		What would you expect the reading on the balance to be after one week?			
			1 mark		
	(ii)	Give a reason for your answer.			
			1 mark		
(b)	(i)	Rust is an oxide of iron. Another oxide of iron is iron(III) oxide. Write a word equation for the formation of iron(III) oxide from its elements.			

(ii) Which one of the following words describes the formation of iron(III) oxide from its elements?

combustion	condensation	decomposition	oxidation

.....

1 mark Maximum 4 marks

3. Some pupils predicted that water will evaporate faster if the surrounding air temperature is higher.To investigate their prediction they placed some water in containers in two different rooms.

(a)	Give two fact	ors they shou	ld keep the s	ame to make	their investigat	ion fair.

1	
	1 mark
2	

(b) They recorded the mass of the water and the container in room 1 and room 2 every day for 5 days.

time	mass of water and container (g)			
(days)	room 1	room 2		
0	100	100		
1	92	85		
2	80	72		
3	72	54		
4	60	45		
5	46	30		

The table below shows their results.

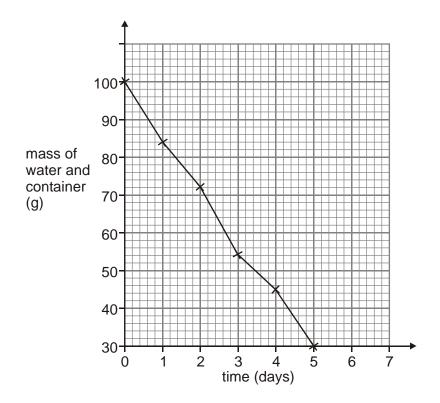
The data shown in their table is **not** sufficient to test their prediction. Explain why.

.....

.....

1 mark

They plotted their data for room 2 and attempted to draw a line of best fit.



(c) Describe the mistake they made in drawing the line of best fit.

1 mark

1 mark

(d) Using the data in the table plot the points for room 1.

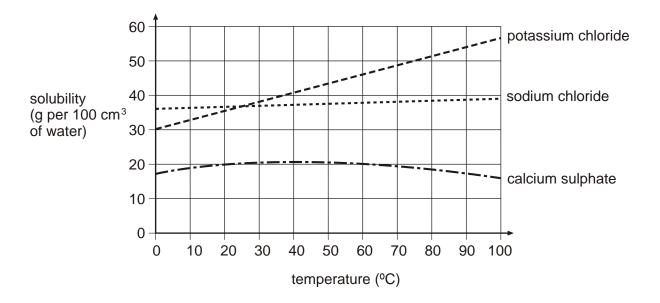
(e) Draw a line of best fit of the points you have drawn.

(f) In which room did the water evaporate more quickly? Tick **one** box.

room 1		room 2	
Use their data to expl	ain your answer.		
	-		

1 mark maximum 7 marks

4. The graph below shows how the solubility of three salts, sodium chloride, potassium chloride and calcium sulphate, changes as the temperature changes.



(a) (i) Use the graph above to compare the solubility of sodium chloride and potassium chloride in the temperature range 10°C to 90°C.



2 marks

(ii)	Ken had a beaker containing 54 g of potassium chloride dissolved in 100 cm ³ of water at 90°C.	
	He cooled the solution to 40°C.	
	What would he see in the beaker as the solution cooled to 40°C?	
	Use the graph to help you.	
		1 mark
	Explain your answer.	
		1 mark

(b) The water in a lake had the three salts dissolved in it. The water evaporated from the lake and the salts were deposited in layers in the order shown below.

potassium chlorid
sodium chloride
calcium sulphate

Look at the graph above.

(i) What evidence is there that these three salts were deposited at a temperature above 25°C?

.....

1 mark

chloride

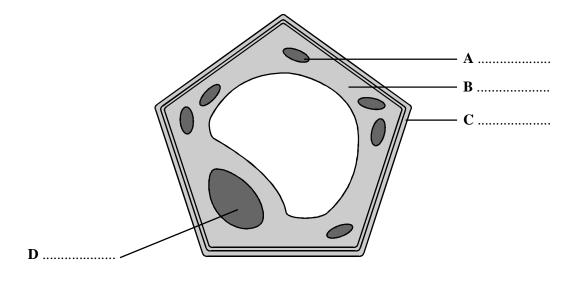
(ii) In what order would the salts be deposited at 10° C?

top
middle
bottom

1 mark maximum 6 marks

BIOLOGY

1. The diagram shows a cell from a blade of grass.



(a) On the diagram, use words from the list to name the parts labelled **A-D.**

cell membrane	cell wall	chloroplast
cytoplasm	nucleus	vacuole

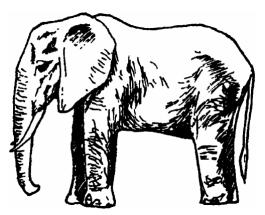
(4)

(b) Name **two** parts of the grass cell which are **not** found in any of the cells in an animal.

1	
-	
2	
	(2)
	(Total 6 marks)

2. The elephant is likely to become extinct in parts of Africa.

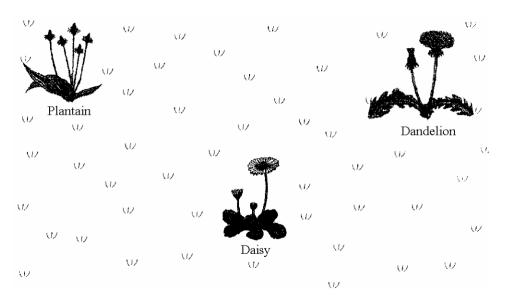
Use the information below to explain three reasons why.



- * The African elephant eats lots of trees and other plants for food.
- * In Africa the human population is increasing and more food is needed to feed the extra people.
- * More trees are cut down for fuel and to clear land for growing crops.
- * Elephants are killed by poachers who want the ivory from their tusks.
- * A herd of elephants needs a large area in which to live and feed.

1	 	 	
2	 	 	
з	 	 	
			(Total 3 marks)

3. The diagram shows three plants on a lawn.



(a) State **one** way in which the leaves of these plants are adapted to survive. Give a reason for your answer.

(2)

(b) Why does grass **not** grow underneath the leaves?

.....

(1) (Total 3 marks)

4. Complete the table to show which part of the blood carries out each function.

Choose your answers from the list.

1	plasma	platelet	red blood cell	white

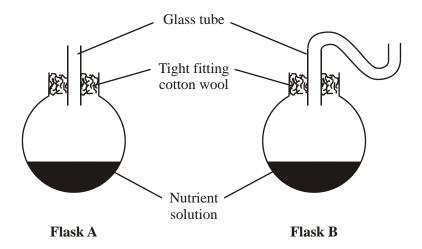
white blood cell

The first answer has been done for you.

Function	Part of the blood
Transports most of the carbon dioxide	plasma
Transports most of the oxygen	
Helps blood to clot at a wound	
Defends the body against microorganisms	
Transports the products of digestion	

(Total 4 marks)

5. In the 1800s, Louis Pasteur investigated the reasons why food goes rotten. To check one of Pasteur's experiments, a student set up the two flasks shown in the diagram.



After three days the nutrient solution in both flasks had gone rotten.

(a) What makes the nutrient solution go rotten?

(1)

(b)	The student then set up two more similar flasks. This time, she boiled the nutrient solution in both flasks for ten minutes.		
	(i)	Why did she boil the nutrient solution?	
			(1)
	(ii)	Why did the nutrient solution in flask A go rotten?	
			(1)
	(iii)	Why did the nutrient solution in flask B not go rotten?	
		(Total 4	(1) marks)

PHYSICS

^{1.} Oliver clamped a wooden plank to a desk. There was a 40 cm overhang as shown in diagram 1.

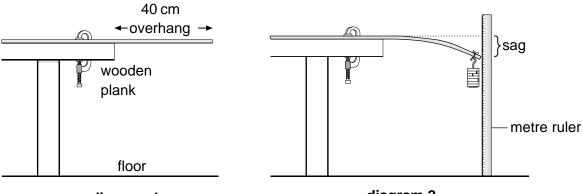
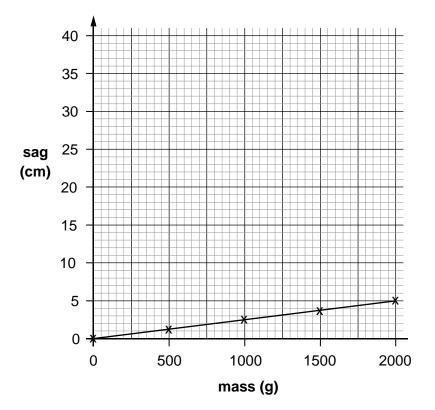




diagram 2

Oliver added masses to the end of the wooden plank as shown in diagram 2. He measured the sag.

The graph below shows his results.



(a) What measurements would Oliver need to take to work out the sag?



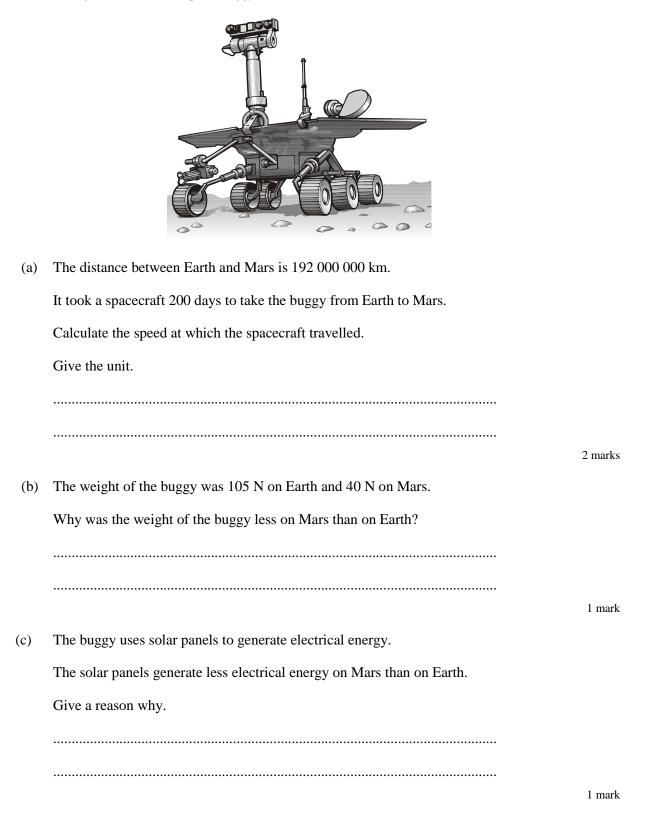
1 mark

(b) Oliver repeated his test with a new plank with an 80 cm overhang. His results are shown below.

mass (g)	sag (cm)
0	1.0
500	15.0
1000	25.0
1500	31.0
2000	35.0

	(i)	Plot the results from Oliver's second test on the grid above. Use the points to draw a line of best fit.	2 marks
	(ii)	In the second test the plank sagged with no mass added to it. Explain what caused this sag.	
			1 mark
(c)	Compare the results of Oliver's two tests.		
	(i)	How are the results similar for each test?	
			1 mark
	(ii)	How are the results different in the second test?	
			1 mark maximum 6 marks

2. The drawing below shows a space buggy on the surface of Mars.



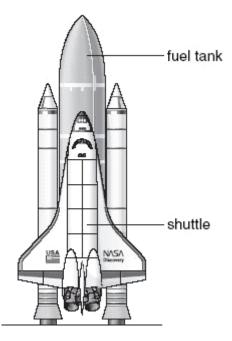
(d) The weight of the buggy was 40 N on Mars.
When the buggy landed on Mars it rested on an area of 0.025 m².

Calculate the pressure exerted by the buggy on the surface of Mars.

Give the unit.

2 marks maximum 6 marks

3. The shuttle is a spacecraft which is used to take satellites into space. The drawing below shows the shuttle just about to take off.



(a) The shuttle has a separate fuel tank containing liquid hydrogen and liquid oxygen.

Explain why hydrogen and oxygen are transported as liquids rather than as gases.

.....

.....

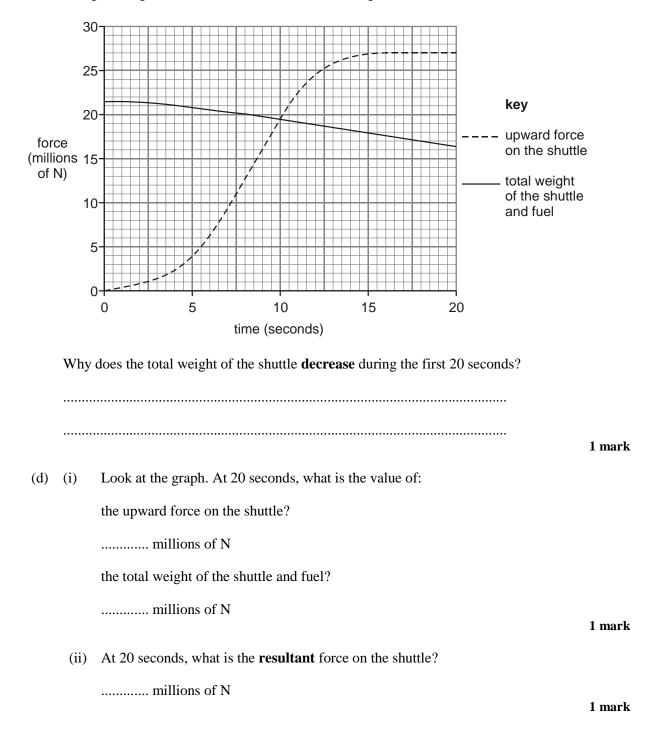
(b) Oxygen is needed to burn the fuel in the shuttle's engines. Vehicles on Earth do **not** need a tank containing oxygen.

Why does the shuttle need to have a tank containing oxygen?

.....

1 mark

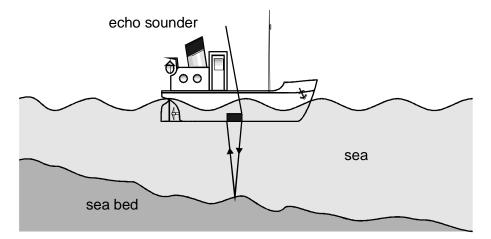
(c) The graph below shows how the upward force and the weight of the shuttle, including fuel, change during the first 20 seconds, after the fuel is ignited.



Use the graph to explain why the shuttle **cannot** take off before 10 seconds. (iii)

..... maximum 6 marks

4. The diagram shows a boat using an echo sounder. It sends a pulse of sound waves which is reflected from the sea bottom. The reflected sound waves are detected by a sensitive microphone.



The time between sending and receiving the pulse is 0.005 s. The device calculates the depth of the sea, using the speed of sound in sea water, which is 1500 m/s.

Calculate the depth of the sea. Show your working. (a)

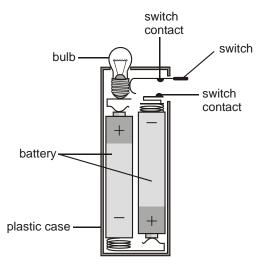
..... 2 marks (b) The boat moves into very deep water. Explain why the reflected pulse is too weak to be detected. 1 mark

(c) The boat's 'echo sounder' could **not** be used in an aeroplane to measure its height above the ground unless it had been modified.

Explain why the device will **not** give correct heights above the ground.

> 1 mark Maximum 4 marks

5. (a) The drawing below shows the parts of a torch.

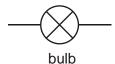


(i) Paul closed the switch. Why did this turn on the torch?



1 mark

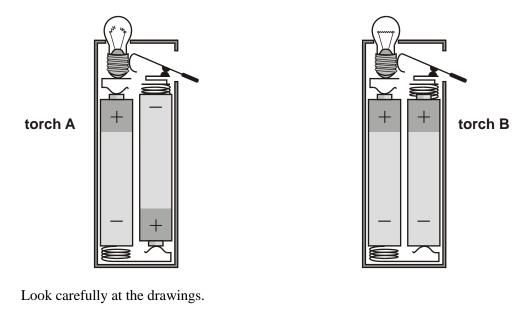
(ii) The diagrams below show symbols for a battery, a bulb and a switch. Connect the symbols to make a series circuit for the torch.



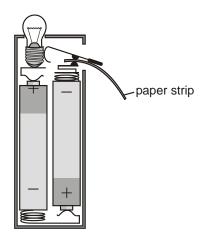
battery

_____switch

(b) The drawings below show two other torches. In both torches, the bulbs will **not** light even when Paul closes the switches.



- (c) When Paul bought his torch there was a paper strip between the contacts of the switch as shown below.



Paul had to remove the paper strip before he could turn the torch on. Give the reason for this.

.....

1 mark maximum 5 marks